

TRIACYL GLICEROLS

Usman Sumo Friend Tambunan

Akli Aditya Parikesit

Bioinformatics Group

Department of Chemistry

Faculty of Mathematics and Science

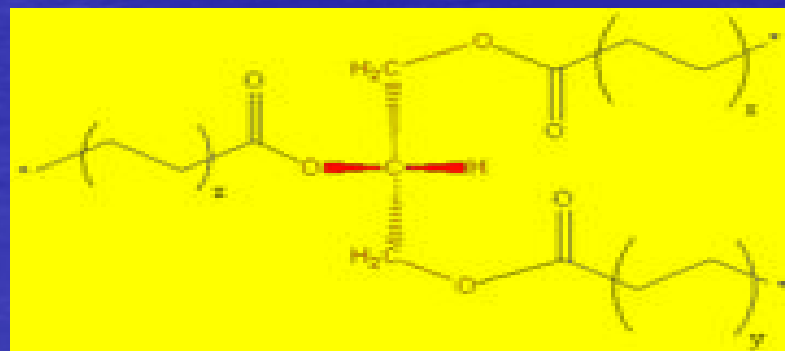
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Energy in Triacylglycerols

- *Triacylglycerols* are highly concentrated stores of metabolic energy because they are *reduced* and *anhydrous*.
- The yield from the complete oxidation of fatty acids is about 9 kcal g⁻¹ (38 kJ g⁻¹), in contrast with about 4 kcal g⁻¹ (17 kJ g⁻¹) for carbohydrates and proteins.
- The basis of this large difference in caloric yield is that fatty acids are much more reduced.

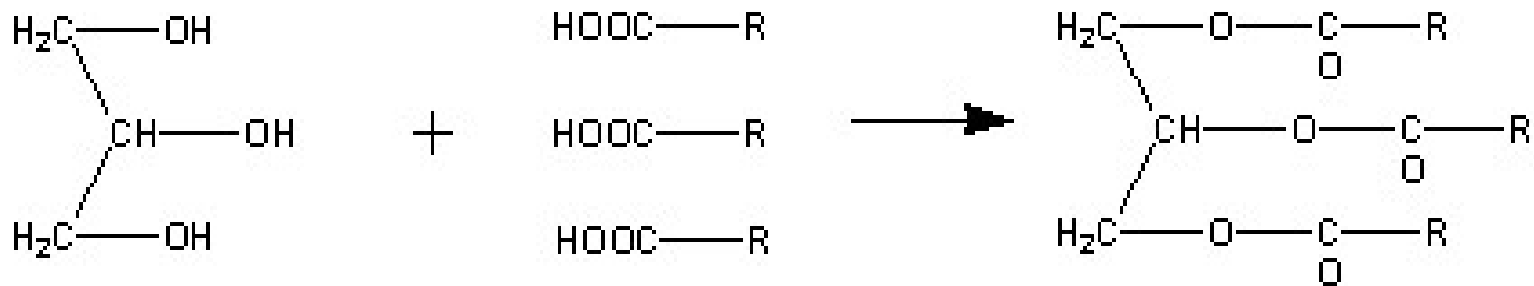
Chemical structure of Triacylglycerol

- The chemical formula is $\text{RCOO-CH}_2\text{CH}(\text{-OOCR}')\text{CH}_2\text{-OOCR''}$, where R, R', and R'' are longer alkyl chains. The three **fatty acids** RCOOH , R'COOH and R''COOH can be all different, all the same, or only two the same.



Biosynthesis of Triacylglycerol

- Esterification process

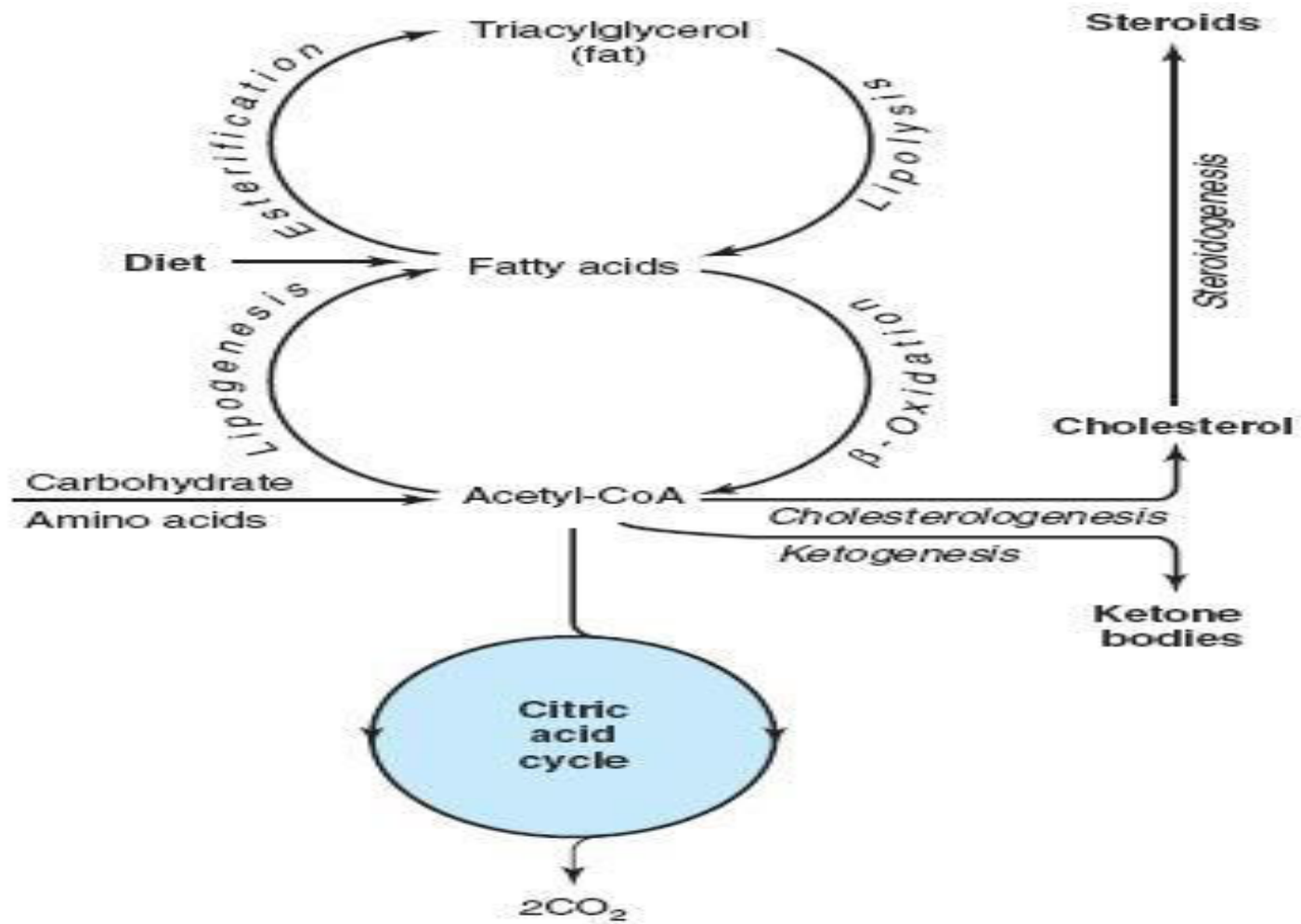


glycerol

fatty acid

TAG

- The major dietary lipid for human, animal and plant are *triacylglycerols*, *sterols*, and *membrane phospholipids*. Fatty acids may be oxidized to acetyl-CoA (β -oxidation) or esterified with glycerol, forming triacylglycerol (fat) as the body's main fuel reserve.

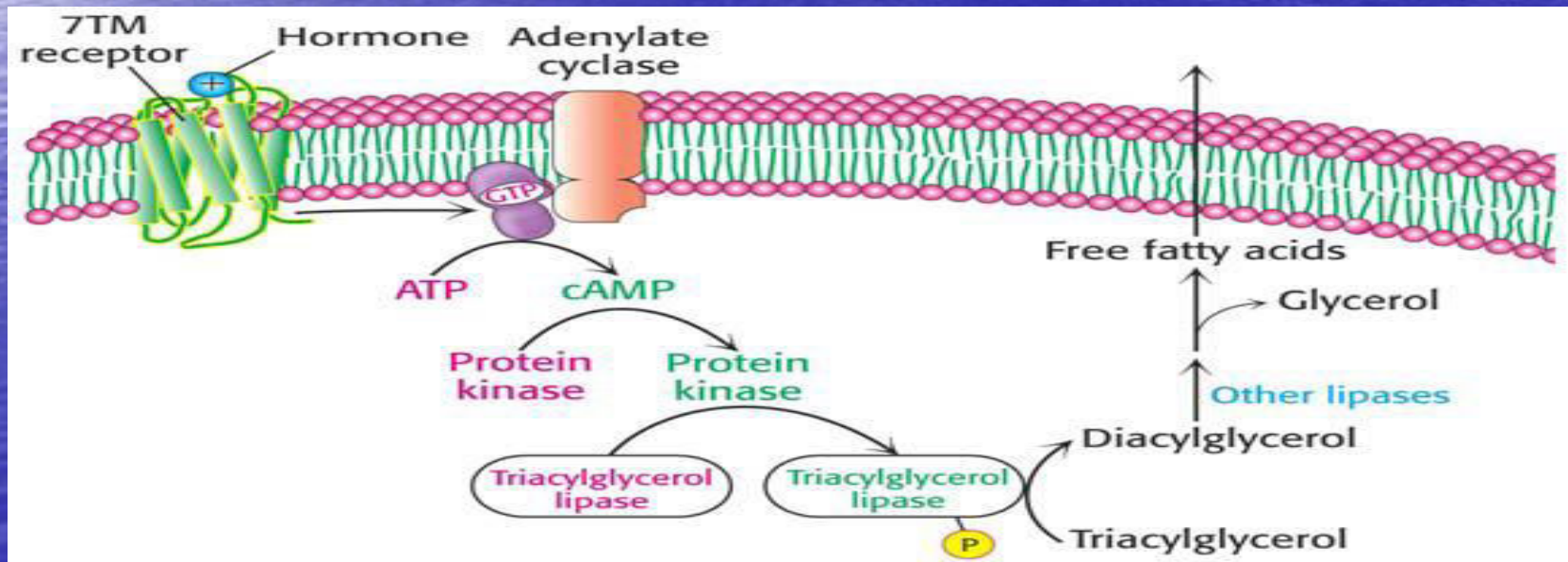


- Furthermore, triacylglycerols are nonpolar, and so they are stored in a nearly anhydrous form, whereas much more polar proteins and carbohydrates are more highly hydrated.
- The glycogen and glucose stores provide enough energy to sustain biological function for about 24 hours, whereas the triacylglycerol stores allow survival for several weeks.

- There are two distinct advantages in storing metabolic energy as fatty acids.
 1. Fatty acids are mainly composed of $\text{—CH}_2\text{—}$ groups which are fully reduced. Therefore, the oxidation of these reduced carbons will yield more energy than oxidized forms of carbon.
 2. Because fatty acids are lipids, they are hydrophobic. They do not need to be solvated in contrast to carbohydrates such as glycogen.

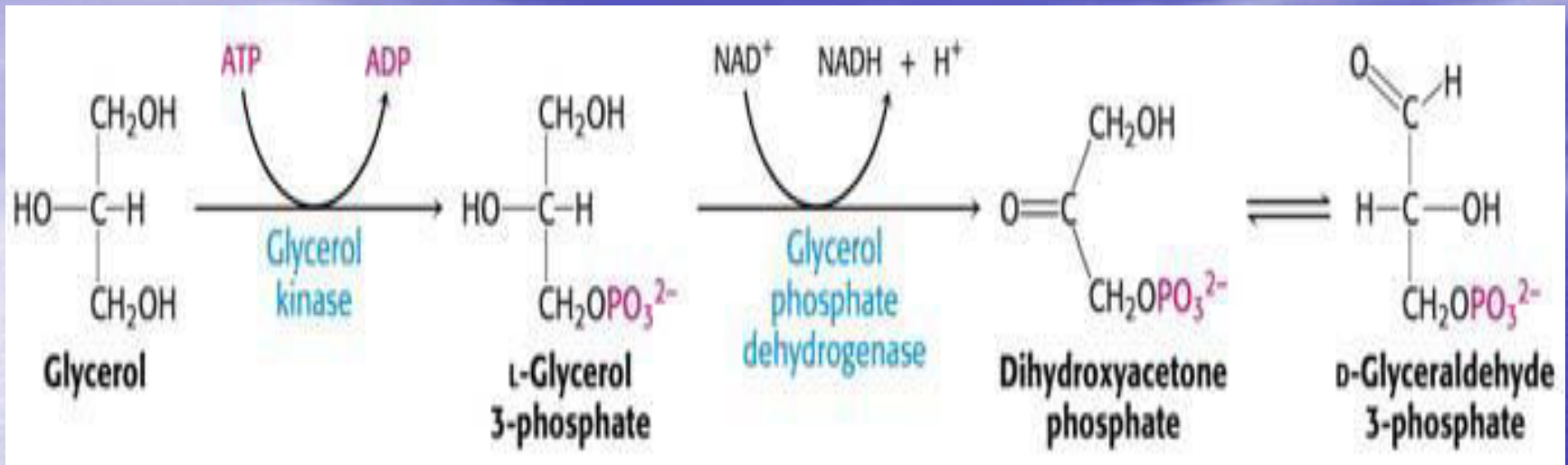
Metabolism of Triacylglycerol in Adipose tissue

- The fatty acids stored in the adipose tissue are mobilized in response to hormone messengers such as epinephrine, norepinephrine, glucagon and adrenocorticotrophic hormone.



Source of Glycerol

- The glycerol formed from lipolysis is absorbed by the liver where it is phosphorylated by **glycerol kinase** to **glycerol-3-phosphate** which is then reduced by **glycerol phosphate dehydrogenase** into **dihydroxyacetone phosphate** which can be converted into **glyceraldehyde 3-phosphate** by **triose phosphate isomerase**.
- The fate of the triose phosphate formed from glycerol can be used in both glycolytic and gluconeogenic pathways depending on the needs of the organism.



Conclusion

- So, the largest amount of biologically available energy in triacylglycerols is in the fatty acid portion, because of its chemical structure.

References

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