# GLYCOLYSIS

Usman Sumo Friend Tambunan Arli Aditya Parikesit Kartika Metafisika

Bioinformatics Grup Department of Chemistry Faculty of Mathematics and Science University of Indonesia



# What is Glycolysis?

- Term: from the Greek glykys, meaning "sweet," And lysis, meaning "splitting"),
- Glycolysis (a sweet splitting process) is a central pathway for the catabolism of carbohydrates in which the six-carbon sugars are split to three-carbon compounds with subsequent release of energy used to transform ADP to ATP. Glycolysis can proceed under anaerobic (without oxygen) and aerobic conditions.
- Some of the free energy released from glucose is conserved in the form of ATP and NADH.

# Glycolisis occur in cytosol in the liver and muscle cells.

Iular Respiration: A Brief Overview:



# Two major step of Glycolisis

#### Preparatory Phases

Collect all heksose chain converted to two molecule of glyceraldehyde 3phosphate

#### payoff phase

Payback the ATP that consumed in preparatory phase

Glyceraldehyde 3-phosphate-> pyruvate



#### **1.** Phosphorylation of Glucose



 $\Delta G^{\prime \circ} = -16.7 \text{ kJ/mol}$ 

#### Conversion of Glucose 6-Phosphate to Fructose 6-Phosphate



 $\Delta G^{\prime \circ} = 1.7 \text{ kJ/mol}$ 

#### Phosphorylation of Fructose 6-Phosphate to Fructose 1,6-





Fructose 1,6-bisphosphate

 $\Delta G'^{\circ} = -14.2 \text{ kJ/mol}$ 

#### Cleavage of Fructose 1,6-Bisphosphate



Fructose 1,6-bisphosphate



 $\Delta G^{\circ} = 23.8 \text{ kJ/mol}$ 

#### Interconversion of the Triose Phosphates

Fructose 1,6-bisphosphate





#### The Payoff Phase of Glycolysis Yields ATP and NADH

#### Oxidation of Glyceraldehyde 3-Phosphate to 1,3-Bisphosphoglycerate





1,3-Bisphosphoglycerate

 $\Delta G^{\prime +} = 6.3 \text{ kJ/mol}$ 

#### Phosphoryl Transfer from 1,3-Bisphosphoglycerate to ADP





Conversion of 3-Phosphoglycerate to 2-Phosphoglycerate



3-Phosphoglycerate

2-Phosphoglycerate

 $\Delta G'^{\circ} = 4.4 \text{ kJ/mol}$ 

#### Dehydration of 2-Phosphoglycerate to Phosphoenolpyruvate



 $\Delta G^{\prime \circ} = 7.5 \text{ kJ/mol}$ 

#### Transfer of the Phosphoryl Group from Phosphoenolpyruvate to ADP



 $\Delta G'^{\circ} = -31.4 \text{ kJ/mol}$ 

#### The Overall Balance Sheet Shows a Net Gain of ATP

Glucose + 2ATP + 2NAD<sup>+</sup> + 4ADP +  $2P_i \rightarrow$ 

2 pyruvate + 2ADP + 2NADH + 2H<sup>+</sup> + 4ATP + 2H<sub>2</sub>O

Canceling out both sides equation:

Glucose + 2NAD<sup>+</sup> + 2ADP +  $2P_i \longrightarrow$ 2 pyruvate + 2NADH +  $2H^+$  + 2ATP +  $2H_2O$ 

#### Case

A compound is an inhibitor of glyceraldehyde-3phosphate.

If this compound were added into liver cells where D-glucose

was only substrate, what is the effect of the additional of this compound for the step of glycolisys?

- In this case, glyceraldehyde-3-phosphate have to continue to the reaction of payoff phase.
- glyceraldehyde-3-phosphate will converted to 1,3biphosphoglycerate as the first compound that produce in the payoff phase
- □ If the enzyme glyceraldehyde 3-phosphate dehydrogenase that has play role in the converting reaction is inhibited, the and the production of 4 molecule of ATP and 2 molecule of NADH will be decrease and the concentration of pyruvate become lower.
- Because the decreasing concentration of pyruvate, the reaction cannot continue to the next step:cytric acid cycle»electron transport chain=the lack of ATP.

### Mechanism of reaction to form glyceraldehyde-3-phosphate to 1,3-biphosphoglycerate



#### Case 2

If the substance were L-lactate, what is the effect of the additional of this compound for the step of glycolisys? L-lactate is usually produced by anaerobic Glycolysis in skeletal muscle return to the liver and converted to glucose, which move back to muscle and converted to glycogen- a circuit called the Cory Cycle.

### Diagram of melabolism



- According to that picture we can see that, L-Lactate can directly converted to pyruvate and going to the next step»citric acid cycle.
- So, the inhibitor of glyceraldehyde 3phosphate do not have an essential effect to the next regulation from glycolysis.

# Referensi

- □ Fry M. 2011, Essential Biochemistry for Medicine Wiley
- Garrett R.H. et al. 2012 Biochemistry Brooks Cole Publishing Company
- Harvey R.A. et al. 2011 Biochemistry Wolters Kluwer Health/Lippincott Williams & Wilkins
- Laemmerhofer M. et al. 2013 Metabolomics in Practice: Successful Strategies to Generate and Analyze Metabolic Data Wiley
- Mathews C.K. et al. 2000 Biochemistry Benjamin Cummings
- Murray R.K. et al. 2012. Harper's Illustrated Biochemistry. 29<sup>th</sup> edition. McGraw Hill Medical.
- Nelson D.L and Cox M.S. 2008. Lehninger Principles of Biochemistry. 5<sup>th</sup> edition. W.H Freeman.

