# Poison of Glycolytic pathway

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The poison include :

- \* sulfhydryl reagent
- \* fluoride
- \* Mercury
- \* Arsenate

#### **Glycolytic Pathway**



## Sulfhydryl Reagent

- Sulfhydryl reagent are usually mercury containing compound or alkylating compound
- Sulfhydryl reagent bring about anhibition at the level of gliceraldehyde-3-phosphate dehydrogenase

 Gliceraldehyde-3-phosphate dehydrogenase catalizes :
Gliceraldehyde-3-p+ NAD+ +pi + 1,3bisphosphateglycerate + NADH+H+ The aldehide of glyceraldehide-3-phosphate reacts with the cystein thiol to form a thiohemiacetal

Enz-cys ——SH





## Enz-cys — S — CH — CH — CH<sub>2</sub>OPO<sub>3</sub><sup>2-</sup> OH OH

#### **Thiohemiacetal Intermediate**

If in the reaction have sulfhydryl reagent, sulfhydryl reagent will react with the sulfhydryl group of glyceraldehyde 3 phosphate to prevent the formation of the thiohemiacetal.

#### $E - Sh + CH_3 - HgCI \rightarrow E - S - Hg - CH_3 + CI^{-1}$

Glyceraldehide 3 Phosphate Dehidrogenase methyl mercury chloride

inaactive enzyme

#### $E - SH + ICH_3CO^{2-} + F - S - CH_2CO_3^{-} + H^+ + I$ iodoacetate Inactive enzime

## **Fluoride Inhibitor**

Enolase catalize 2-Phosphoglycerate Phosphoenolpyruvate + H<sub>2</sub>O



This Mg2+ dependent dehydration reaction is inhibite by fluride

Mg2+ and inorganik phosphate are belived to form an ionik complex with fluride

### **Arsenate Inhibitor**



 The arsenate is look like inorganic phosphate and is able to substitute the phosphate in enzime catalized reaction

D-glycerakdehide-3-phosphate + HAsO42-+NAD+ → 1-arsenato-3-phospho-Dglycerate + NADH + H+

## 1-arsenato-3-phospho-D-glycerate is unstable

1-arsenato-3-phospho-D-glycerate +  $H_2O$   $\longrightarrow$  3-Phospho-D-glycerate +  $HAsO_4^{2-}$ +  $H^+$ 

The presence of arsenate, 1,3bisphosphate is not formed, resulting in the loss capacity to synthesize ATP

- The consequence is that net ATP synthesis does not occur when glycolysis happens
- The presence of arsenate, glycolysis does not generate ATP, which can be used to meet the energy of a cell

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